

LETTERS  
TO THE EDITORReactions of Spiro[2.3]hexane-5-carbonyl Chloride  
with Phosphorus(III) Acids Esters

Yu. N. Mitrasov, D. A. Sosnov, and O. V. Kondrat'yeva

Yakovlev Chuvash State Pedagogical University, ul. K. Marksa 38, Cheboksary, 428000 Russia

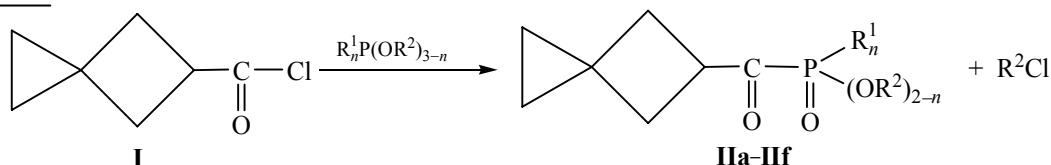
e-mail: sosnovd1987@mail.ru

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We have previously shown that the reaction of bicyclo[4.1.0]heptane-7-carbonyl chloride with medium esters of phosphorous and arylphosphonous acids occurs with the retention of the small carbocycle [1]. In continuation of this work it is interesting to study the behavior of carbonyl compounds of spirohexane series in the Arbuzov reaction [2]. Due to the high energy strain of these spiranes one could expect unusual reaction course affording new types of organophosphorus compounds.

We found that spiro[2.3]hexane-5-carbonyl chloride **I** reacts readily with phosphorous and arylphosphonous acids esters. When an equimolar ratio of reagents was used under mild conditions (0–20°C), the attack of the P-nucleophiles occurs on the carbon atom of the carbonyl group. As a result, the reaction produces dialkyl spiro[2.3]hexane-5-carbonyl phosphonates **IIa–IIc** or alkylarylspiro[2.3]hexane-5-carbonyl phosphinates **IId–IIf**.



$n = 0$ ,  $R^2 = C_2H_5$  (**a**),  $C_3H_7$  (**b**),  $C_4H_9$  (**c**);  $n = 1$ ,  $R^1 = C_6H_5$ ,  $R^2 = C_2H_5$  (**d**),  $C_3H_7$  (**e**),  $C_4H_9$  (**f**).

The structure of compounds **IIa–IIf** was confirmed by the IR and  $^1H$  NMR spectroscopy. Thus, in the IR spectra there are low-intensive peaks of the stretching vibrations of the C–H bonds of the three-membered ring in the range of 3090–3095  $cm^{-1}$  and the strong absorption bands of carbonyl, phosphoryl, and ester groups at 1700–1705, 1255–1260, 995–1050  $cm^{-1}$ , respectively. The  $^1H$  NMR spectra the three-membered ring protons resonate in a strong field ( $\delta$  0.61 ppm). The four-membered ring protons were observed in a weaker field as a doublet at  $\delta$  2.77 ppm (4H,  $CH_2$ ,  $^3J_{HH}$  7.75 Hz) and a multiplet at 3.24 ppm (1H, CH). The protons of alkoxy groups and aromatic ring appear in characteristic areas.

**Compound IIa.** Yield 75%, bp 151–153°C (5 mm Hg),  $d_4^{20}$  1.1693  $g\ cm^{-3}$ ,  $n_D^{20}$  1.4774.

**Compound IIb.** Yield 82%, bp 161–162°C (2 mm Hg),  $d_4^{20}$  1.1307  $g\ cm^{-3}$ ,  $n_D^{20}$  1.4765.

**Compound IIc.** Yield 76%, bp 167–168°C (1 mm Hg),  $d_D^{20}$  1.1006  $g\ cm^{-3}$ ,  $n_D^{20}$  1.4752.

**Compound II d.** Yield 81%, bp 182–183°C (1 mm Hg),  $d_4^{20}$  1.1992  $g\ cm^{-3}$ ,  $n_D^{20}$  1.5456.

**Compound IIe.** Yield 78%, oily substance,  $d_4^{20}$  1.1792  $g\ cm^{-3}$ ,  $n_D^{20}$  1.5418.

**Compound II f.** Yield 79%, oily substance,  $d_4^{20}$  1.1594  $g\ cm^{-3}$ ,  $n_D^{20}$  1.5378.

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